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The units of matter are called atoms. An atom is the smallest possible amount of element that retains that element's properties. Each atom has at its center a dense positively charged nucleus, which is surrounded at some distance by a cloud of negatively charged electrons. Electrons held in orbit by electrostatic attraction to the nucleus.

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THE PROPERTIES AND STRUCTURE OF MATTER . COURSE CONTENT . 1. Define matter and state of matter 2. Properties of solids, liquids and gases 3. Changes in matter ... "Extremely small chemically indivisible particle" Atom is Greek for "that which cannot be divided"

### **THE PROPERTIES AND STRUCTURE OF MATTER - NUST**

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### **A Small Matter of Programming | The MIT Press**

The three most common phases of matter on Earth are solids, liquids and gases. 3) Identify and describe the particle in each phase of matter and how they are different in each phase of matter.

### **Phases of Matter – Multiple Choice Quiz**

2 Unit 3 Matter matter liquid solid gas change materia lquido slido gas cambio evaporation boil freeze burn rust ... atoms, called molecules, help us understand the properties of matter. A small number of elements can join to form many different combinations. LESSON 5 Matter Changes in Appearance

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### **States of matter - John Wiley & Sons**

Matter is made up of particles too small to be seen. The three states of matter are solid, liquid, and gas. Solids, liquids and gases can be characterized by their properties (including volume, shape, and the movement and spacing of particles). Different densities of water, or solutions, will stratify to form layers.

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