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### Organic Chemistry: Jonathan Clayden, Nick Greeves, Stuart

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In a blatant plug for the Reagent Guide, each Friday I profile a different reagent that is commonly encountered in Org 1/ Org 2. Version 1.2 just got released, with a host of corrections and a new page index. Having just talked about the oxidation ladder, it makes sense to start going into reagents for oxidation and reduction reactions.. Sodium borohydride (NaBH<sub>4</sub>)

### Sodium Borohydride: Reducing Agent for aldehydes and

Going for gold: The mono- and di-chloroform solvates of [Au<sub>2</sub>(μ<sub>2</sub>-1,2-bis(diphenylarsino)ethane)<sub>2</sub>](AsF<sub>6</sub>)<sub>2</sub> undergo single-crystal-to-single-crystal transformations that result in gain or loss of only one chloroform molecule and changes in the structure and luminescence of the digold cation.

### Chemistry – A European Journal: Early View

Properties. Phenol is an organic compound appreciably soluble in water, with about 84.2 g dissolving in 1000 mL (0.895 M). Homogeneous mixtures of phenol and water at phenol to water mass ratios of ~2.6 and higher are possible.

### Phenol - Wikipedia

Acetic acid / É™ Èˆ s iÈ• t Éª k /, systematically named ethanoic acid / Èœ È› Î, È™ Èˆ n oÈŠ Éª k /, is a colourless liquid organic compound with the chemical formula CH<sub>3</sub>COOH (also written as CH<sub>3</sub>CO<sub>2</sub>H or C<sub>2</sub>H<sub>4</sub>O<sub>2</sub>). When undiluted, it is sometimes called glacial acetic acid. Vinegar is no less than 4% acetic acid by volume, making acetic acid the main component of vinegar apart from water.

### Acetic acid - Wikipedia

Advances in Carbocation Chemistry . I sotopes in Organic Chemistry . Advances in Carbohydrate Chem. and Biochem. Molecular Structure and Energetic

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