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TEXT: Physical Chemistry, by Ira N. Levine, McGraw Hill, 6th Edition. This text is available in a hardcover edition and in a paperback (and less expensive) edition.

Chemistry 75â€™ General Information - Dartmouth College

Physical Chemistry Problems. Â©Mike Lyons 2013. JF Physical Chemistry 2013-2014. JF CH 1101: Introduction to Physical Chemistry . Professor Mike Lyons. ... NaOH. (i) What is the pH of the solution when 90 mL of 0.1 M NaOH has been added to 100 mL of 0.1 M acetic acid. (ii) Determine the pH at the equivalence point. (iii) What is the pH

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Abstract: Recognizes that modern physical chemistry is based on quantum mechanics and that it makes pedagogical sense to begin with the atomic and molecular perspective and use it to build a firm understanding of macroscopic phenomena.

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